

## **THIOCYANATE (SCN<sup>-</sup>) ION TEST KIT**

### **OFI Part No. 144-94**

Thiocyanate ion (SCN<sup>-</sup>) is used as a tracer in water base drilling fluids. This test kit provides all of the necessary supplies and reagents to determine the amount of Thiocyanate ion in a drill stem test fluid or mud filtrate. For colored filtrates it is necessary to also prepare a blank sample which is helpful in determining the color of the end point.

#### **Equipment:**

- #153-14 Cylinder, Graduated, 50 x 1 ml, glass
- #153-34 Pipette, 1 ml x 1/100 ml, glass
- #153-36 Pipette, 2 ml x 1/10 ml, glass
- #153-38 Pipette, 5 ml x 1/10 ml, glass
- #153-50-1 Flask, Erlenmeyer, 125 ml, glass
- #153-60 Syringe, Disposable, 3 ml
- #154-75 Scoop, Brass

#### **Reagents:**

- #145-551 Starch Indicator sol'n, 2 oz
- #206-01 Distilled Water, 8 oz
- #144-941 \*Bromine Water, 8 oz (UN1744)
- #144-942 \*Orthophosphoric Acid, 20% sol'n, 8 oz (UN1805)
- #144-943 \*Phenol solution, 5%, 8 oz (UN2821)
- #144-944 Potassium Iodide Crystals, 50 grams
- #262-05 Sodium Thiosulfate solution, 0.01N, 8 oz

#### **Case:**

- #144-35 Case, stainless steel, diagonal design
- #163-28 Clip, large

**Procedure:**

1. Pipette 5 ml of filtrate into a 125 ml Erlenmeyer Flask and add about 45 ml of Distilled Water.
2. Pipette 5 ml of 20% Orthophosphoric acid solution and mix by swirling the contents.
3. Add Bromine water dropwise while mixing until a deep yellow color persists and allow the mixture to stand for 5 minutes.
4. Add 2 ml of 5% Phenol solution and mix until the deep yellow color disappears.  
*Note: Add all of the 5% Phenol solution at once.*
5. Add one full scoop of Potassium Iodide crystals and mix until dissolved.
6. Allow the mixture to stand for 5 minutes.
7. Add 10 - 20 drops of Starch Indicator solution.
8. Titrate with 0.01N Sodium Thiosulfate solution to a colorless end point.  
*Note: Colored filtrates require the preparation of a blank sample to determine the color of the end point. Omit steps 5 - 8 of the procedure for the blank sample and titrate the test sample to the same end point color as the color of the blank sample after step 4.*

**Calculation:**

$$\text{Thiocyanate Ion (SCN}^{-}\text{)} = \text{Sodium Thiocyanate, ml} \times 58.1$$

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